Application of 3D NMR for Structure Determination of Peptide Natural Products

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S Supporting Information

[AB](#page-5-0)STRACT: [Despite the a](#page-5-0)dvances in NMR, structure determination is often slow and constitutes a bottleneck in natural products discovery. Removal of this bottleneck would greatly improve the throughput for antibiotic discovery as well as other therapeutic areas. Overall, faster structure methods for structure determination will serve the natural products community in a broad manner. This report describes the first application of 3D NMR for elucidation of two microbially produced peptide natural products with novel structures. The methods are cost-effective and greatly improve the confidence in a proposed structure.

■ **INTRODUCTION**

Drug resistant infectious diseases, in particular Gram-negative infections, continue to threaten global health as well as contemporary medical practices.¹ For example, as pathogens become increasingly resistant, semiroutine procedures such as organ transplants and joint [r](#page-5-0)eplacements will become increasingly dangerous. Combined with the emergence of pan-resistant pathogens, new antibiotics are sorely needed. Although natural products have provided the antibiotic basis upon which many medical practices were built, discovering novel antibiotics has become more challenging. Nonetheless, advances in sequencing and genomics has not only indicated that there is more potential than previously thought, but also have provided tools for nearly routine analyses of the genetic potential of bacteria to produce small molecules, many of which would have antibiotic potential. $2,3$ In parallel, analytical strategies that take advantage of increasing sensitivity of mass spectrometers as well as the spe[ed](#page-5-0) and resolving power of UPLC systems have enabled rapid identification of novel antibiotics.3−¹² Combining these new tools with the bacterial diversity found in underexplored ecological niches such as insect sym[b](#page-5-0)i[on](#page-5-0)ts, 13 thermal vents, 14 caves, 15 coal mines, 16 and marine ecosystems 17 has demonstrated great promise for discovering the [nex](#page-5-0)t generation o[f a](#page-5-0)ntibio[tic](#page-5-0) therapies.

To address the c[ha](#page-5-0)llenges of resistant infectious disease, the potential of these genetic and analytical methods must be fully realized. Our lab has focused on LCMS-based metabolomics as a discovery platform to discover novel natural products from marine invertebrate-associated bacteria. This has led to a high discovery rate of novel natural products.^{8−11} As these methods

pave the way for rapid discovery, structure determination becomes a limiting factor for providing valuable lead structures. Therefore, we have developed methods such as isotopic labeling followed by ¹³C−¹³C gCOSY^{11,18,19} for rapid assembly of carbon backbones in polyketides and ¹³C−¹⁵N NMR¹¹ to assist with thiazolines and other he[terocyc](#page-5-0)lic systems. While our previous studies demonstrated a high level of ^{13}C ^{13}C ^{13}C enrichment could be achieved for molecules of acetate origin, questions remained surrounding isotopically labeling peptides. For example, typical growth media used were rich in natural abundance precursors such as amino acids from sources like yeast extract and peptone. On the other hand, if uniform labeling could be achieved for both 13 C and 15 N, structure determination of peptide-based natural products could be achieved using biomolecular NMR strategies such as 3D NMR techniques. These approaches would also provide a gateway to automated backbone assignment of structures containing typical amino acids. Therefore, we evaluated isotopic labeling in a medium that contained typical quantities of nonenriched components. To demonstrate proof of concept for both isotopic labeling and 3D NMR strategies, we tested the methods using two novel antibiotic peptides, one of which was prioritized as part of an antibiotic drug discovery program. In particular, peptide 2 showed broad spectrum activity including against Gram-negative pathogens. Although peptide 2 did not show high potency, the difference between the MIC (22.7 μ M) and MBC (45.5 μ M) was small indicating potential for Gram-

Received: June 30, 2015 Published: August 14, 2015 negative infections. The overall goal of this study was to provide a basis for reducing structure determination times from months to hours or days.

■ RESULTS AND DISCUSSION

Uniform Labeling Peptides with $13C$ and $15N$. Our previous studies showed that fermentation of actinobacteria upon addition of uniformly 13 C-labeled glucose and subsequent purification yielded ¹³C-labeled natural products.¹⁹ However, peptides could be more challenging since in most cases U⁻¹³C glucose was utilized for 13 C enrichment in nat[ura](#page-5-0)l products which have been biosynthesized via a polyketide pathway.^{11,20} For the ^{15}N incorporation, $^{15}NH_4Cl$ was evaluated as the nitrogen source. As mentioned, a potential issue was ¹⁵N incorporation with peptides in rich media. To evaluate the ${}^{13}C$ and ¹⁵N incorporation, a marine invertebrate-derived Streptomyces sp. (strain WMMB 705) that produced two novel peptides that we named eudistamides $A(1)$ and $B(2)$ was grown in multiple media and extracts were analyzed by LC−MS data (Supporting Information). Although the medium containing 15NH4Cl as the only nitrogen source (M3, Supporting Infor[mation\) yielded the h](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)ighest ¹⁵N abundance of the peptides, taking into consideration the various fact[ors such as](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [the price of](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) the medium, the yield, and the 13 C and 15 N abundance, the labeled medium containing peptone and yeast extract (20 g soluble starch, 10 g $[U-$ ¹³C] glucose, 2.5 g peptone, 2.5 g yeast extract, 5 g $^{15}NH_4Cl$, 5 g $CaCO_3/L$ of artificial seawater) was used for producing 13 C- and 15 N-labeled peptides.

Strategies and Experiments for the Resonance Assignment of Uniformly 13 C, 15 N-Labeled Peptides. Backbone Assignment Strategy. A typical triple resonance 3D NMR spectrum observes correlations of three resonances, a proton, a carbon, and a nitrogen. Many 3D NMR experiments have been designed for assigning the backbone of $^{15}N/^{13}C$ isotopically labeled protein, such as $HNCO₁²¹NH(CA)CO₂²²$ $HNCA₁²¹$ CBCANH₁²³ CBCA(CO)NH₁²⁴ HN(CO)CA₁²⁵ HCACO,²¹ HCA(CO)N,²¹ H(CCO)NH,^{[26](#page-6-0)} C(CO)NH,²⁶ and HB[HA](#page-6-0)(CBCACO[\)N](#page-6-0)H 27 Ideally, stand[ard](#page-6-0) triple resonan[ce](#page-6-0) backbone [as](#page-6-0)signment of a [pep](#page-6-0)tide has relied [on](#page-6-0) a combinati[on](#page-6-0) of the CBCANH, CBCA(C[O\)](#page-6-0)NH, and NHCO spectra (Figure 1a−c). The general idea was that the CBCANH correlated each NH group with the C_{α} and C_{β} chemical shifts of its own residue (strongly) and of the preceding residue (weakly), whereas the CBCA(CO)NH only correlates the NH group to the preceding C_{α} and C_{β} chemical shifts. The NHCO experiment provides the connectives between each NH group with the carbonyl carbon of the preceding residue, thus establishing the backbone assignment. In our experience, the quality of the CBCANH and CBCA(CO)NH spectra was sometimes not sufficient in terms

Figure 1. Connectives observed in seven different types of 3D NMR experiment. (a) The CBCANH experiment correlates the chemical shift of amide resonance with the C_{α} and C_{β} of both inter- and intraresidue. (b) The CBCA(CO)NH experiment only correlates the amide resonance with C_{α} and C_{β} of the preceding residue. (c) The NHCO experiment provides the connectives between the amide of a residue (NHⁱ) with the carbonyl carbon of the preceding residue (CO_{i-1}) . (d) The NH(CA)CO experiment shows the amide resonance (NH_i) is correlated with the carbonyl carbon of the same residue (CO_i) , as well as that of the preceding residue (CO_{i-1}) . (e) The NHCA experiment correlates the amide resonance (NH_i) with the C_a of the same residue and preceding residue. (f) The NH(CO)CA experiment only correlates the resonances of the amide with the C_{α} of the preceding residue. (g) The HCCH-TOCSY correlates a ${}^{1}H/{}^{13}C$ pair to all other protons in the same aliphatic side chain. (h) The thick arrow represents pathways utilizing the 1 J (C^{a/ δ} , N) coupling, whereas the thin arrow represents the less efficient pathway via ^{2}J (C^a, N). Protons directly involved in the transfer pathways are shown in bold font.

of S/N. The C_{α} or C_{β} resonances were, for example, not visible above the noise level. In this case, the combination of HNCA and HN(CO)CA experiments (Figure 1e−f) which were more sensitive provided the same information as the CBCANH and $CBCA(CO)NH$ spectra, except without the C_β resonances. The HN(CA)CO experiment (Figure 1d) was powerful for linking each NH group with the carbonyl carbon of the same residue, particularly helpful to the assignment of depsipeptides. A depsipeptide is a peptide in which one or more of its −C(O)NHR groups are replaced by the corresponding ester, −C(O)OR. Taking eudistamides A (1) and B (2) for example, no correlation for C-1 could be seen in NHCO spectrum, whereas HN(CA)CO experiment could correlate NH-7 with

the carbonyl carbon of the ester (C-1). Therefore, our backbone assignment strategy used three 3D NMR experiments, HNCO, CBCANH, and CBCA(CO)NH, to establish the connectivity of the backbone as the first step. The other 3D NMR experiments, NHCA, HN(CO)CA, NH(CA)CO, were utilized to make further confirmation regarding to some complicated cases, such as overlapping carbonyl carbon signals and any crosspeak that was very weak or missing in either CBCANH or CBCA(CO)NH spectrum. We were thereby able to make unambiguous backbone assignment of the eudistamides A and B and provided a general strategy for non-Pro containing peptide natural products. However, due to the absence of an amide proton in proline, these 3D NMR experiments cannot be used to sequentially connect proline residue to the preceding residue. The CDCA(NCO)CAHA (Figure 1h) was designed for the sequential assignment of proline residues, 28 and the other option is to rely on the [observatio](#page-1-0)n of NOEs to neighboring residues.

Side Chain [As](#page-6-0)signment Strategy. Depending on the structure, side chains can be assigned using TOCSY and HSQC. However, the most useful NMR experiment for side chain assignment is an HCCH-TOCSY spectrum (Figure 1g).²⁹ The general principle behind using the HCCH-TOCSY spectrum is as follows: Using the known C_{α} and C_{β} c[hemical](#page-1-0) [sh](#page-1-0)if[ts](#page-6-0) from the backbone assignment, the side chain chemical shifts would be assigned by viewing 2D slices at each carbon shift, typically at the C_{α} and C_{β} chemical shifts. Each "slice" contains resonances for the protons attached to carbons in that ¹³C−¹³C spin system. The HCCH-TOCSY experiment combined with 13C chemical shift ranges of the common amino acids provides a very powerful means for the assignment of the aliphatic side chain. Some of the proteinogenic amino acids, such as alanine, serine, threonine and glycine, were easily identified since alanine, serine, and threonine's C_β chemical shifts were unique compared to those of the other amino acids and glycine has no C_{β} . Valine, isoleucine, and proline could also be easily identified because they have lower than normal C_{α} chemical shifts. For the aromatic side chain assignment, such as tyrosine, phenylalanine, tryptophan, and histidine, additional 2D NMR data are needed to complete the whole side chain assignment.

Structure Elucidation of Eudistamides A (1) and B (2). Eudistamide A (1) was obtained as a white powder with a molecular formula of $C_{61}H_{90}N_{14}O_{13}$, determined by HRESIMS. Interpretation of HCCH-TOCSY data (Table S1; Supporting Information) of the 13 C- and 15 N-doubly labeled 1 established the amino acid residues leucine (Leu; 2×), argi[nine \(Arg\),](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [proline \(Pro](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)), glycine (Gly), glutamine (Gln), alanine (Ala), valine (Val), and phenylalanine (Phe), which were further confirmed by natural abundance experiments, ¹H-¹H COSY, HSQC and HMBC data (Table S3; Supporting Information). The Pro residue was assigned as *cis* based on the 13 C NMR chemical shifts of the *β*- and *γ*-carbo[n atoms \(](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) $\Delta \delta_{\beta\gamma}$). $\Delta \delta_{\beta\gamma}$ for *trans*-Pro is regularly less than 5 ppm, while $\Delta \delta_{\beta\gamma}$ for *cis*-Pro is regularly between 5 and 10 ppm. 30 $\Delta \delta_{\beta\gamma}$ for the proline residue in 1 was 5.8 ppm, supporting the assignment as cis. The 3-(2 methylphenyl)acrylic acid (Me-a[cyl](#page-6-0)) moiety was established by HMBC, whereas geometry of the double bond (C_{66} and C_{67}) was assigned as E on the basis of large coupling constant (15.5 Hz) through the double bond.

The connectivity of the partial structures $Thr-Phe-Leu₂-Val-$ Ala-Gln-Gly and Pro-Arg-Leu₁ was established by using 3D NMR data which were discussed for the backbone assignment (CBCANH, CBCACONH, NHCO, HNCA, NH(CO)CA, $NH(CA)CO$; Table S1; Supporting Information) of the ¹³Cand 15N-doubly labeled 1. The HMBC correlation from H-62 to C-1 and the ROESY correlation between H_3 -63 and H-2 connected Leu₁ to Thr via an ester on the basis of the chemical shift of C-62 (δ c 68.5). Moreover, for the uncommon amino acid residue, the HMBC correlations from H-66 and H-67 to C-65 supported the linkage of Me-acyl moiety to Thr. The ROESY correlations between H-24a and $H₂$ -22 secure the ring closure by connecting the Gly unit to Pro, which satisfied the 24 degrees of unsaturation deduced from the molecular formula.

The advanced Marfey's method 31 was applied to assign the absolute configurations of the amino acid residues from acid hydrolysis of 1. The 1-fluoro-2,4-d[ini](#page-6-0)trophenyl-5-leucine-amide (FDLA) derivatives of the hydrolysate of 1 and authentic Dand L-amino acids were subjected to LC−MS analysis. The absolute configuration of all amino acids was established by comparison of their HPLC retention time and molecular weights with those of corresponding authentic D- and Lstandards. Upon analysis, the amino acid residues Arg, Pro, Ala, Val, allo-Thr, and Phe were determined to have the Lconfiguration, whereas the Gln and Leu were deduced to have the D-configuration (Table S5; Supporting Information).

Eudistamide B (2) was isolated a white amorphous powder with a molecular formula of $C_{72}H_{99}N_{11}O_{18}$ as determined by HRESIMS. A detailed analysis of H[CCH-TOCSY](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [\(Table](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [S](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)2; Supporting Information) of the 13 C- and 15 N-doubly labeled 2 indicated an amino acid content as follows: Thr $(x2)$, Tyr $(x2)$, Ser, Lys, 3-OH-Leu $(x2)$, Leu, and Phe. CBCANH, CBCA(CO)NH NHCO, HNCA, NH(CO)CA, and NH(CA)- CO spectra (Table S2; Supporting Information) revealed the amino acids sequence as 3 -OH-Leu₁-Thr₁-Tyr₁-Ser-3-OH-Leu₂Lys-Tyr₂-Leu-Phe-Thr₂, [which was further c](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)onfirmed by ESIMS/MS fragmentation analysis (Figure S22; Supporting Information). The linkage of Me-acyl moiety to Thr_2 was supported by the HMBC correlation from H-76 to [C-74. The](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [ROESY corr](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)elation between H_3 -72 and H-2 and consideration of unsaturation deduced from the molecular formula connected 3-OH-Leu₁ to Thr₂ via an ester on the basis of the chemical shift of C-71 (δ c 68.5).

The absolute configurations of the amino acids were determined using acid hydrolysis followed by the advanced Marfey's method. The chromatographic comparison between Marfey's derivatives of the hydrolysate of 2 and appropriate amino acid standards assigned the L-configurations for Ser and Leu and D-configurations for Lys, Phe, allo-Thr, and Tyr (Table S4; Supporting Information). The (2R, 3S) and (2R, 3R)-3 hydroxyleucine were synthesized following Bonnard's procedure³² [and derivatized with](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) L- and L/D-FDLA respectively. Since enantiomers would exhibit identical retention behavior und[er](#page-6-0) nonchiral HPLC conditions, the (2R, 3S)-D-FDLA derivative showed the same retention time as (2S, 3R)-L-FDLA, whereas the $(2R, 3R)$ -D-FDLA derivative showed the same retention time as (2S, 3S)-L-FDLA. By comparing the above the four retention times of the 3-OH-Leu standards with the L-FDLA derivatives of the hydrolysate of 2, both the 3-OH-Leu in 2 were assigned as 2S, 3R (Table S6; Supporting Information). The 2S, 3R configuration of the two 3 hydroxyleucines in 2 were also confirmed by J-ba[sed analysis](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [and NOESY](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) data. The large coupling constants observed from DQF-COSY in MeOD between H-2 and H-3 (11.5 Hz), and between H-28 and H-29 (11.0 Hz) established their antirelationship. The ROESY correlation between H-4 and NH-7, as well as the ROESY correlation between H-30 and NH-33, established the rotamer (A4) in Figure 2.

Figure 2. J-based configuration analysis of C-2 and C-3 of compound 2.

Compounds 1 and 2 were tested for antibacterial activity against Pseudomonas aeruginosa, Methicillin-resistant Staphylococcus aureus (MRSA), Escherichia coli, and Bacillus subtilis. Compound 2 showed antibacterial activity against MRSA, E. coli and B. subtilis, with the MIC values of 22.7, 22.7, and 2.8 μ M, respectively, while compound 1 did not show any detectable antibacterial activities at 52.2 μ M. Compound 2 was also tested using a murine thigh model infected with either E. coli ATCC 25922 or S. aureus ATCC 25923 (Figure 3). Over the 16-fold intraperitoneal dose range, no toxicity associated with compound administration was observed (Figure 3a,c). Conversely, intravenous dose levels above 8 mg/kg resulted in animal death (Figure 3b,d). Modest, but broad spectrum (both

Figure 3. In vivo time kill experiment with compound 2 using a neutropenic mouse thigh model. Each symbol represents the mean and standard deviation from four thighs of two mice infected with either E. coli or S. aureus. The error bars represent the standard deviation. Five single doses (4, 8, 16, 32, and 64 mg/kg) of compound 2 were administered to mice. (a) E. coli infected murine thighs treated with five single doses of 2 by intraperitoneal route. (b) E. coli infected murine thighs treated with five single doses of 2 by intravenous route. (c) S. aureus infected murine thighs treated with five single doses of 2 by intraperitoneal route. (d) S. aureus infected murine thighs treated with five single doses of 2 by intravenous route. The (b) and (d) regimens do not include the 16, 32, and 64 mg/kg since the mice died following administration.

Gram-positive and Gram-negative bacteria) efficacy was observed with two highest tolerated dose levels given by both administration routes. Compared to untreated control animals, compound 2 reduced the organism burden in the thighs of immunocompromised mice by more than 0.5 log_{10} cfu/thigh.

Although the cyclopetides and cyclodepsipeptides have been encountered frequently as bacterial secondary metabolites, eudistamides $A(1)$ and $B(2)$ differ markedly from the known peptides by virtue of the presence of a unique 3-(2 methylphenyl)acrylic acid moiety, which has not been previous reported in a peptide. Only some analogues of 3-(2 methylphenyl)acrylic acid residue were rarely descried in peptides, such as $WS9326s$, 33 pepticinnamins, 34 and mohangamides³⁵ incorporating a 2-pentenyl cinnamic acid moiety, skyllamycins A and B36,37 i[nco](#page-6-0)rporating a 2-pr[op](#page-6-0)enyl]-cinnamic acid [mo](#page-6-0)iety, and corprisamides A and B bearing 2-heptatrienyl cinnamic acid moiet[y.](#page-6-0)³⁸ [M](#page-6-0)oreover, another remarkable feature of compound 2 is the high content of the β -hydroxy- α -amino acids, including gene[tic](#page-6-0)ally coded proteinogenic amino acids Ser and Thr, and nonstandard amino residue 3-OH-Leu. Members of the β -hydroxy- α -amino acid class occur as constituents of many important antibiotics, such as vancomycin³⁹ and lysobacin.⁴⁰ Furthermore, compounds 1 and 2 could be produced in sufficient quantities by scale-up microbial fer[me](#page-6-0)ntation, whic[h w](#page-6-0)ould greatly facilitate follow-up mechanistic and preclinical studies.

■ CONCLUSION

The introduction of 3D NMR has dramatically improved the speed and reliability of the protein assignment process.^{22,41} However, we expanded the scope of 3D NMR experiments from protein applications to a strategy that is univer[sally](#page-6-0) applicable for labeled peptides. We have demonstrated a rapid and efficient structure elucidation protocol for complex peptides containing typical amino acids. The protocol we propose consists of a typical set of 3D NMR experiments for backbone and side chain assignment of uniformly labeled peptides with 13 C and 15 N. While 3D NMR experiments may not be optimal for natural product peptides, this proof of concept provides a basis for using triple resonance strategies for rapidly establishing the backbone of peptide natural products. Development of specific NMR experiments would be advantageous. The triple resonance experiments used in this study were highly complementary to more traditional methods such as HMBC and ROESY. This report describes the first application of 3D NMR to two microbially produced peptide natural products with novel structures. The methods are cost-effective and greatly improve the confidence in a proposed structure.

EXPERIMENTAL SECTION

General Experimental Procedures. Optical rotations were measured on a Polarimeter. UV spectra were recorded on a UV−vis Spectrophotometer. IR spectra were measured with a FT−IR Spectrophotometer. 1D and 2D NMR spectra were obtained in DMSO with a NMR spectrometer equipped with a 13 C/¹⁵N{¹H} cryoprobe. 3D NMR spectra were obtained in DMSO with a NMR spectrometer with a $({}^{1}H\{{}^{13}C,{}^{15}N\})$ cold probe. HRMS data were acquired with a QTOF mass spectrometer (Ionization method: Spray Voltage (+) 4000; Spray Voltage (−) 3700; Capillary temperature 320; full MS scan 200−2000). MSMS data were acquired with a hybrid quadrupole-Orbitrap mass spectrometer. RP HPLC was performed using a HPLC system and a Phenomenex Luna C_{18} column (250 \times 10 mm, 5μ m), as well as a preparative HPLC and Phenomenex Gemini

 C_{18} column (250 × 30 mm, 5 μ m). The Advanced Marfey's method utilized a HPLC coupled with a mass spectrometer.

Biological Material. Ascidian specimens were collected in September 2011 from the Florida Keys (24°33.416′, 81°21.611′). Identification was confirmed by Shirley Parker-Nance. A voucher specimen for Eudistoma olivaceum (Van Name, 1902)⁴² is housed at the University of Wisconsin-Madison. For cultivation, a sample of ascidian $(1\,\,\mathrm{cm}^3)$ $(1\,\,\mathrm{cm}^3)$ $(1\,\,\mathrm{cm}^3)$ was rinsed with sterile seawater and macerated using a sterile pestle in a microcentrifuge tube, and dilutions were made in sterile seawater, with vortexing between steps to separate bacteria from heavier tissues. Dilutions were separately plated on three media: ISP2
supplemented with artificial seawater,⁴³ R2A,⁴⁴ and M4.⁴⁵ Each medium was supplemented with 50 $\mu\rm g/mL$ cycloheximide and 25 $\mu\rm g/$ mL nalidixic acid. Plates were incubate[d at](#page-6-0) 28 °[C f](#page-6-0)or at least [28](#page-6-0) days, and strain WMMB 705 was purified from an ISP2 isolation plate.

Sequencing. Genomic DNA was extracted using the UltraClean Microbial DNA Isolation kit (Mo Bio Laboratories, Inc.). 16S rDNA genes were amplified using 100−200 ng genomic DNA template with the primers 8−27F (5′ to 3′ GAGTTTGATCCTGGCTCAG) and 1492R (5′ to 3′ GGTTACCTTGTTACGACTT). The following PCR conditions were used: 94 °C for 5 min, followed by 30 cycles of 94 °C for 30 s, 55 \degree C for 1 min, 72 \degree C for 1.5 min, with a final step of 72 \degree C for 5 min. The PCR bands were excised from the gel and purified using the QIAquick Gel Extraction kit (QIAGEN). One microliter of purified product was sequenced. Sequencing reactions were performed by the UW Biotechnology Center and reactions were sequenced with an ABI 3730xl DNA Analyzer. WMMB 705 were identified as Streptomyces sp. by 16S sequencing, and demonstrated 99% sequence similarity to Streptomyces sp. FXJ6.141 (accession number GU 339421).

Fermentation, Extraction, and Isolation. Two 10 mL of seed cultures (25×150 mm tubes) in medium ASW-A (20 g soluble starch, 10 g glucose, 5 g peptone, 5 g yeast extract, 5 g $CaCO₃$ per liter of artificial seawater) were inoculated with strain WMMB 705 and shaken (200 rpm, 28 °C) for 4 days. Two-liter flasks (2 \times 500 mL) containing ASW-A medium with Diaion HP20 (7% by weight) were inoculated with 8 mL from the culture tube and shaken at 200 rpm and 28 °C for 7 days. Filtered HP20 was washed with water and extracted with acetone. The acetone extract was chromatographed on Diaion HP20ss, eluting sequentially with methanol/water (10:90 and 95:5). The 95% methanol eluate was evaporated to dryness to afford a yellow oil (460 mg), and then it was subjected to RP HPLC (30%/70% to 60%/40% $ACN/H₂O$ with $H₂O$ containing 0.1% acetic acid over 30 min, 25 mL/ min, followed by 90/10% to 100% of the same solvents for 3 min, 25 mL/min, and a hold at 100%/0% of the same solvents) using a Phenomenex Gemini C₁₈ column (250 × 30 mm, 5 μ m), yielding a fraction containing 1 (50 mg, t_R 11.5 min), and compound 2 (21 mg, t_R 12.5 min). The fraction containing 1 was further subjected to RP HPLC (55-65% MeOH-H₂O with H₂O containing 0.1% acetic acid over 25 min, 4.0 mg/mL) using a Phenomenex Luna C_{18} column (250 \times 10 mm, 5 μ m), yielding 1 (22 mg, t_R 19 min). For ¹³C and ¹⁵N incorporation, the same procedure was used $(1 \times 250 \text{ mL})$ with labeled medium ASW-A (20 g soluble starch, 10 g $U^{13}C$ -glucose, 2.5 g peptone, 2.5 g yeast extract, 5 g $^{15}NH_4Cl$, 5 g CaCO₃ per liter of artificial seawater).

Antibacterial Assay. Eudistamides $A(1)$ and $B(2)$ were tested for antibacterial activity against MRSA (ATCC #33591), E. coli (ATCC #25922), B. subtilis, and P. aeruginosa (ATCC #27853), and MICs were determined using a dilution antimicrobial susceptibility test for aerobic bacteria.⁴⁶ Eudistamides A (1) and B (2) were dissolved in DMSO, serially diluted to 10 concentrations (0.125−64 μg/mL), and tested in a 96-wel[l p](#page-6-0)late. Vancomycin was used as a control and exhibited an MIC of 0.69 μ mol against MRSA and 0.69 μ mol against B. Subtilis. Gentamicin was used as a control and exhibited an MIC of 8.4 μmol against E. coli and 8.4 μmol against P. aeruginosa. Peptides 1 and 2 were tested in triplicate, and vancomycin and gentamicin were also tested in triplicate. Six untreated media controls were included on each plate. The plates were incubated at 35 °C for 18 h. The MIC was determined as the lowest concentration that inhibited visible growth of bacteria.

Eudistamide A (1). White solid, $[\alpha]^{25}$ _D −51 (c 0.1, MeOH); UV (MeOH) λ (log ε) 209 (3.55), 224 (3.17), 280 (3.23); IR (ATR) v_{max} 3300, 2953, 1648, 1535, 1519, 1450, 1320, 1242, 1072, 760, 677 cm⁻¹;
¹H and ¹³C NMR (see Table S3; Supporting Information); HRMS [M ¹H and ¹³C NMR (see Table S3; Supporting Information); HRMS [M + H]⁺ m/z 1227.6882 (calcd for C₆₁H₉₁N₁₄O₁₃, 1227.6890).

Eudistamide B (2). White solid, $[\alpha]^{25}$ _D + 37 (c 0.1, MeOH); UV (MeOH) λ (log ε) 207 (3.32), 2[23](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [\(3.17\),](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [279](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [\(2.82\);](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [IR](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) (ATR) v_{max} 3290, 2932, 1738, 1647, 1532, 1517, 1455, 1314, 1244, 1072, 761, 669 cm⁻¹; ¹H and ¹³C NMR (see Table S4; Supporting Information); HRMS $[M + H]^+$ m/z 1406.7253 (calcd for $C_{72}H_{100}N_{11}O_{18}$, 1406.7248).

Acid Hydrolysis of Eudistamides A (1[\)](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [and](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [B](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf) [\(](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)2). Separate solutions of compounds 1 and 2 (0.5 mg each) in 6 N HCl (1 mL) were hydrolyzed at 110 °C for 4 h and dried under vacuum.

Determination of Proteinogenic Amino Acid Configura**tions.** L - and L/D -FDLA were synthesized as previously reported.⁴ The hydrolysate was mixed with 1 N NaHCO₃ (40 μ L), and 35 μ L of L-FDLA (10 mg/mL in acetone). Each solution was stirred at 45 $^{\circ}$ [C](#page-6-0) for 1 h, cooled to room temperature, quenched with 1 N HCl (40 μ L), and dried under vacuum. Similarly, the standard L- and D-amino acid were derivatized separately. The derivatives of the hydrolysate of compound 1 and the standard amino acids were subjected to LC−MS analysis with a Phenomenex Kinetex C_{18} reversed-phase column (2.6) μ m, 100 × 4.6 mm) at a flow rate of 0.5 mL/min and with a linear gradient of H_2O (containing 0.1% formic acid) and MeOH (90:10 to 0:100 over 15 min, and a hold at 100% MeOH for 5 min). The absolute configurations of the amino acids were determined by comparing the retention times of the L- and D-amino acids derivatives, which were identified by MS. The absolute configurations of the amino acids in 1 were assigned based on a comparison of retention time of amino acid standards derivatized with L-FDLA. In a similar manner, the derivatives of the hydrolysate of compound 2 and the standard amino acids were subjected to LC−MS analysis with a Phenomenex Luna C₁₈ reversed-phase column (5.0 μ m, 250 \times 4.6 mm) at a flow rate of 1.0 mL/min and with a linear gradient of H_2O (containing 0.1% formic acid) and ACN (90:10 to 0:100 over 30 min, and a hold at 100% ACN for 5 min). The retention times and ESIMS data for L-FDLA derivatives of the hydrolysates and the standard amino acids are summarized in Tables S5 and S6 (Supporting Information).

Synthesis of (2R,3S)-3-Hydroxyleucine and (2R,3R)-3-Hydroxyleucine and Advanced Marfey's Analysis of 3-Hydroxyleucines. (2R,3S)-3-Hydroxyleuc[ine and \(2](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)R,3R)-3-hydroxyleucine were synthesized using the procedures by Bonnard et al.³² We obtained 20.0 mg of major precursor (S)-1-((2R,5S)-5-isopropyl-3,6 dimethoxy-2,5-dihydropyrazin-2-yl)-2-methylpropan-1-ol and [2.5](#page-6-0) mg of minor precursor (R)-1-((2R,5S)-5-isopropyl-3,6-dimethoxy-2,5 dihydropyrazin-2-yl)-2-methylpropan-1-ol. The hydroxyleucine precursors were hydrolyzed following the general peptide acid hydrolysis process. The major precursor gave a mixture of (2R,3S)-3 hydroxyleucine and D-valine, and the minor precursor gave a mixture of (2R,3R)-3-hydroxyleucine and D-valine. These two hydrolysate mixtures were derivatized with L-FDLA and L/D-FDLA, respectively, resulting in four products: (2R, 3S)-3-hydroxyleucine-L-FDLA, (2R, 3S)-3-hydroxyleucine-D-FDLA, (2R, 3R)-3-hydroxyleucine-L-FDLA, and (2R, 3R)-3-hydroxyleucine-D-FDLA. Among these four products, (2R, 3S)-3-hydroxyleucine-D-FDLA will show the same retention time as (2S, 3R)-3-hydroxyleucine-L-FDLA, and (2R, 3R)-3-hydroxyleucine-D-FDLA will show the same retention time as (2S, 3S)-3 hydroxyleucine-L-FDLA, under the nonchiral HPLC condition we used in advanced Marfey's analysis. The resulting four derivatization mixtures were subjected to LC−MS analysis with a Phenomenex Luna C_{18} reversed-phase column (5.0 μ m, 250 \times 4.6 mm) at a flow rate of 1.0 mL/min and with a linear gradient of H_2O (containing 0.1%) formic acid) and ACN (25:75 to 65:35 over 50 min). When these four retention times were compared with the retention time of hydrolysate-L-FDLA derivative of compound 2, the two 3-hydroxyleucines in the peptide were assigned as 2S, 3R. Detailed retention times of four isomers and hydrolysate could be found in Table S6 (Supporting Information). ¹H NMR data of (S) -1- $((2R,5S)$ -5-isopropyl-3,6dimethoxy-2,5-dihydropyrazin-2-yl)-2-methylpropan-1-ol (CDCl₃,

500 MHz): δ 4.09 (1H, m), 3.99 (1H, m), 3.73 (3H, s), 3.69 (3H, s), 3.63 (1H, br), 2.25 (1H, m), 2.01, (1H, m), 1.64 (1H, br), 1.04 (3H, d, $J = 7.0$ Hz), 1.03 (3H, d, $J = 7.0$ Hz), 0.99 (3H, d, $J = 7.0$ Hz), 0.71 (3H, d, J = 7.0 Hz). ¹H NMR data of (R) -1- $((2R,5S)$ -5-isopropyl-3,6dimethoxy-2,5-dihydropyrazin-2-yl)-2-methylpropan-1-ol $(CDCl₃, 500$ MHz): δ 4.10 (1H, t, J = 3.5 Hz), 3.92 (1H, t, J = 3.5 Hz), 3.65 (3H, s), 3.64 (3H, s), 3.59 (1H, br), 2.20 (1H, m), 1.76, (1H, m), 1.51 (1H, br), 0.98 (3H, d, J = 7.0 Hz), 0.85 (3H, d, J = 7.0 Hz), 0.84 (3H, d, J = 7.0 Hz), 0.63 (3H, d, $J = 6.9$ Hz).

Organisms, Media, and Antibiotic of in Vivo Studies. Eudistamide B (2) was tested for antibacterial activity in vivo studies. One isolate of E. coli ATCC 25922 and S. aureus ATCC 25923, separeatly, was used for these studies. Organisms were grown, subcultured, and quantified using Mueller-Hinton broth (MHB) and agar (Difco Laboratories, Detroit, MI).

Animals. Six week-old, specific pathogen-free, female ICR/Swiss mice weighing 24−27 g were used for all studies (Harlan Sprague− Dawley, Indianapolis, IN). Animals were maintained in accordance with the American Association for Accreditation of Laboratory Animal Care (AAALAC) criteria. All animal studies were approved by the Animal Research Committees of the William S. Middleton Memorial VA Hospital and the University of Wisconsin.

Murine Thigh Infection Model. Mice were rendered neutropenic (neutrophils <100/mm³) by injecting cyclophosphamide (Mead Johnson Pharmaceuticals, Evansville, IN) intraperitoneally 4 days (150 mg/kg) and 1 day (100 mg/kg) before thigh infection. Previous studies have shown that this regimen produces neutropenia in this model for 5 days. Broth cultures of freshly plated bacteria were grown to logarithmic phase overnight to an absorbance of 0.3 at 580 nm (Spectronic 88; Bausch and Lomb, Rochester, NY). After a 1:10 dilution into fresh MHB, bacterial counts of the inoculum ranged from 106.3 to 6.9 cfu/mL. Thigh infections with each of the isolates were produced by injection of 0.1 mL of inoculum into the thighs of isoflorane-anesthetized mice 2 h before therapy.

In Vivo Time Kill and PAE. Two hours after thigh infection with either single doses of compound 2 $(4, 8, 16, 32, \text{ and } 64 \text{ mg/kg})$ by the intraperitoneal or intravenous route, groups of two treated and untreated mice were sacrificed at each treatment end point (0 h at the start of therapy and 6 h after treatment). The thighs (four per treatment group) were immediately removed upon euthanasia and processed for cfu determination. The burden of organisms in the thigh was measured by viable plate counts of tissue homogenates. The impact of each dose on the burden of organisms was measure over time of the study period.

■ ASSOCIATED CONTENT

6 Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b01486.

NMR data, MS data, and Marfey's analysis for [compounds](http://pubs.acs.org) 1 and 2. (PDF)

■ AUTHOR INFORMATI[ON](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01486/suppl_file/jo5b01486_si_001.pdf)

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Notes

The authors declare no competing fi[nancial interest](mailto:tim.bugni@wisc.edu).

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■ REFERENCES

(1) Livermore, D. M. Clin. Microbiol. Infect. 2004, 10, 1−9.

(2) (a) Lautru, S.; Deeth, R. J.; Bailey, L. M.; Challis, G. L. Nat. Chem. Biol. 2005, 1, 265−269. (b) Bentley, S. D.; et al. Nature 2002, 417, 141−147.

(3) Doroghazi, J. R.; Albright, J. C.; Goering, A. W.; Ju, K.; Haines, R. R.; Tchalukov, K. A.; Labeda, D. P.; Kelleher, N. L.; Metcalf, W. M. Nat. Chem. Biol. 2014, 10, 963−968.

(4) Forner, D.; Berrué, F.; Correa, H.; Duncan, K.; Kerr, R. G. *Anal.* Chim. Acta 2013, 805, 70−79.

(5) Gill, K. A.; Berrue, F.; Arens, J. C.; Kerr, R. G. ́ J. Nat. Prod. 2014, 77, 1372−1376.

(6) Hoffmann, T.; Krug, D.; Hüttel, S.; Müller, R. Anal. Chem. 2014, 86, 10780−10788.

(7) Nguyen, D. D.; Wu, C.; Moree, W. J.; Lamsa, A. L.; Medema, M. H.; Zhao, X.; Gavilan, R. G.; Aparicio, M.; Atencio, L.; Jackson, C.; Ballesteros, J.; Sanchez, J.; Watrous, J. D.; Phelan, V. V.; van de Wiel, C.; Kersten, R. D.; Mehnaz, S.; Mot, R. D.; Shank, E. A.; Charusanti, P.; Nagarajan, H.; Duggan, B. M.; Moore, B. S.; Bandeira, N.; Palsson, B. Ø.; Pogliano, K.; Gutiérrez, M.; Dorrestein, P. C. Proc. Natl. Acad. Sci. U. S. A. 2013, 110, E2611−E2620.

(8) Hou, Y.; Braun, D. R.; Michel, C. R.; Klassen, J. L.; Adnani, N.; Wyche, T. P.; Bugni, T. S. Anal. Chem. 2012, 84, 4277−4283.

(9) Wyche, T. P.; Hou, Y.; Vazquez-Rivera, E.; Braun, D.; Bugni, T. S. J. Nat. Prod. 2012, 75, 735−740.

(10) Hou, Y.; Tianero, D. B.; Kwan, J. C.; Wyche, T. P.; Michel, C. R.; Ellis, G. A.; Vazquez-Rivera, E.; Braun, D. R.; Rose, W. E.; Schmidt, E. W.; Bugni, T. S. Org. Lett. 2012, 14, 5050−5053.

(11) Wyche, T. P.; Piotrowski, J. S.; Hou, Y.; Braun, D.; Deshpande, R.; McIlwain, S.; Ong, I. M.; Myers, C. L.; Guzei, I. A.; Westler, W. M.; Andes, D. R.; Bugni, T. S. Angew. Chem., Int. Ed. 2014, 53, 11583− 11586.

(12) Samat, N.; Tan, P. J.; Shaari, K.; Abas, F.; Lee, H. B. Anal. Chem. 2014, 86, 1324−1331.

(13) (a) Kikuchi, Y. Microbes Environ. 2009, 24, 195−204. (b) Poulsen, M.; Oh, D.; Clardy, J.; Currie, C. R. PLoS One 2011, 6, e16763.

(14) (a) Sogin, M. L.; Morrison, H. G.; Huber, J. A.; Welch, D. M.; Huse, S. M.; Neal, P. R.; Arrieta, J. M.; Herndl, G. J. Proc. Natl. Acad. Sci. U. S. A. 2006, 103, 12115−12120. (b) Andrianasolo, E. H.; Haramaty, L.; Rosario-Passapera, R.; Bidle, K.; White, E.; Vetriani, C.; Falkowski, P.; Lutz, R. J. Nat. Prod. 2009, 72, 1216−1219.

(15) (a) Schabereiter-Gurtner, C.; Saiz-Jimenez, C.; Pinar, G.; Lubitz, W.; Rölleke, S. FEMS Microbiol. Ecol. 2003, 1606, 1−13. (b) Engel, A. S.; Porter, M. L.; Stern, L. A.; Quinlan, S.; Bennett, P. C. FEMS Microbiol. Ecol. 2004, 51, 31−53.

(16) (a) Wang, X.; Shaaban, K. A.; Elshahawi, S. I.; Ponomareva, L. V.; Sunkara, M.; Zhang, Y.; Copley, G. C.; Hower, J. C.; Morris, A. J.; Kharel, M. K.; Thorson, J. S. J. Nat. Prod. 2013, 76, 1441−1447. (b) Shaaban, K. A.; Wang, X.; Elshahawi, S. I.; Ponomareva, L. V.; Sunkara, M.; Copley, G. C.; Hower, J. C.; Morris, A. J.; Kharel, M. K.; Thorson, J. S. J. Nat. Prod. 2013, 76, 1619−1626.

(17) (a) Lam, K. S. Curr. Opin. Microbiol. 2006, 9, 245−251. (b) Abdelmohsen, U. R.; Bayer, K.; Hentschel, U. Nat. Prod. Rep. 2014, 31, 381−399.

(18) Reibarkh, M.; Wyche, T. P.; Bugni, T. S.; Martin, G. E.; Williamson, R. T. Structure elucidation of uniformly ¹³C labeled small molecule natural products. Magn. Reson. Chem. 2015, in press.

(19) Ellis, G. A.; Wyche, T. P.; Fry, C. G.; Braun, D. R.; Bugni, T. S. Mar. Drugs 2014, 12, 1013−1022.

(20) (a) Fellermeier, M.; Eisenreich, W.; Bacher, A.; Zenk, M. H. Eur. J. Biochem. 2001, 268, 1596−1604. (b) Watanabe, H.; Tetsuo, T.; Oikawa, H. Tetrahedron Lett. 2006, 47, 1399−1402. (c) Kwon, Y.; Park, S.; Shin, J. Arch. Pharmacal Res. 2014, 37, 967−971.

(21) Kay, L. E.; Ikura, M.; Tschudin, R.; Bax, A. J. Magn. Reson. 1990 , 89, 496 −514.

(22) Clubb, R. T.; Thanabal, V.; Wagner, G. J. Magn. Reson. 1992 , 97, 213 −217.

(23) Grzesiek, S.; Bax, A. J. Magn. Reson. 1992 , 99, 201 −207.

(24) Grzesiek, S.; Bax, A. J. Am. Chem. Soc. 1992, 114, 6291–6293. (25) Bax, A.; Ikura, M. J. Biomol. NMR 1991, 1, 99-104. ,

,

(26) Grzesiek, S.; Anglister, J.; Bax, A. J. Magn. Reson., Ser. B 1993

101, 114 −119. (27) Grzesiek, S. J.; Bax, A. J. Biomol. NMR 1993 3, 185 −204. ,

(28) Bottomley, M. J.; Macias, M. J.; Liu, Z.; Sattler, M. J. Biomol. NMR 1999 , 13, 381 −385.

(29) Bax, A.; Clore, M.; Gronenborn, A. M. J. Magn. Reson. 1990 , 88 , 425 −431.

(30) Siemion, I. Z.; Wieland, T.; Pook, K. H. Angew. Chem., Int. Ed. Engl. 1975, 14, 702-703.

(31) (a) Harada, K.; Fujii, K.; Mayumi, T.; Hibino, Y.; Suzuki, M. Tetrahedron Lett. 1995 , 36, 1515 −1518. (b) Harada, K.; Fujii, K.; Hayashi, K.; Suzuki, M. M. Tetrahedron Lett. 1996, 37, 3001–3004.

(c) Fujii, K.; Ikai, Y.; Oka, H.; Suzuki, M.; Harada, K. Anal. Chem. 1997 , 69, 5146 −5151.

(32) Bonnard, I.; Rolland, M.; Salmon, J.; Debiton, E.; Bartheomeuf, C.; Banaigs, B. J. Med. Chem. 2007, 50, 1266-1279.

(33) (a) Hayashi, K.; Hashimoto, M.; Shigematsu, N.; Nishikawa, M.; Ezaki, M.; Yamashita, M.; Kiyoto, S.; Okuhara, M.; Kohsaka, M.; Imanaka, H. J. Antibiot. 1992 , 45, 1055 −1063. (b) Yu, Z.; Vodanovic-Jankovic, S.; Kron, M.; Shen, B. Org. Lett. 2012, 14, 4946−4949.

(34) Shiomi, K.; Yang, H.; Inokoshi, J.; Van der Pyl, D.; Nakagawa, A.; Takeshima, H.; Omura, S. J. Antibiot. 1993 , 46, 229 −234.

(35) Bae, M.; Kim, H.; Moon, K.; Nam, S.; Shin, J.; Oh, K.; Oh, D. Org. Lett. 2015, 17, 712-715.

(36) Toki, S.; Agatsuma, T.; Ochiai, K.; Saitoh, Y.; Ando, K.; Nakanishi, S.; Lokker, N. A.; Giese, N.; Matsuda, Y. J. Antibiot. 2001 , 54, 405 −414.

(37) Pohle, S.; Appelt, C.; Roux, M.; Fieldler, H.; Sü ssmuth, R. D. J. Am. Chem. Soc. 2011 , 133, 6194 −6205.

(38) Um, S.; Park, S. H.; Kim, J.; Park, H. J.; Ko, K.; Bang, H.; Lee, S. K.; Shin, J.; Oh, D. Org. Lett. 2015, 17, 1272-1275.

(39) (a) Wohlleben, W.; Stegmann, E.; Sü ssmuth, R. D. Methods Enzymol. 2009 , 458, 459 −486. (b) Gunasekera, S. P.; Ritson-Williams, R.; Paul, V. J. J. *Nat. Prod.* **2008**, 71, 2060–2063. (c) Cai, G.; Napolitano, J. G.; McAlpine, J. B.; Wang, Y.; Jaki, B. U.; Suh, J.; Yang, S. H.; Lee, I.; Franzblau, S. G.; Pauli, G. F.; Cho, S. J. Nat. Prod. 2013 , 76, 2009 −2018. (d) Hou, Y.; Tianero, D. B.; Kwan, J. C.; Wyche, T. P.; Michel, C. R.; Ellis, G. A.; Vazquez-Rivera, E.; Braun, D. R.; Rose, W. E.; Schmidt, E. W.; Bugni, T. S. Org. Lett. 2012, 14, 5050-5053. (40) O 'Sullivan, J.; McCullough, J. E.; Tymiak, A. A.; Kirsch, D. R.; Trejo, W. H.; Principe, P. A. J. Antibiot. 1988, 41, 1740−1744. (41) (a) Salzmann, M.; Pervushin, K.; Wider, G.; Senn, H.; Wüthrich,

K. J. Am. Chem. Soc. 2000 , 122, 7543 −7548. (b) Liang, B.; Tamm, L. K. Proc. Natl. Acad. Sci. U. S. A. 2007 , 104, 16140 −16145. (c) Sakakibara, D.; Sasaki, A.; Ikeya, T.; Hamatsu, J.; Hanashima, T.; Mishima, M.; Yoshimasu, M.; Hayashi, N.; Mikawa, T.; Walchli, M.; ̈ Smith, B. O.; Shirakawa, M.; Güntert, P.; Ito, Y. Nature 2009, 458 , 102 −106.

(42) Van Name, W. G. Z. Kristallogr. - Cryst. Mater. 1902, 11, 325-412.

(43) Harrison, P. J.; Waters, R. E.; Taylor, F. J. R. J. Phycol. 1980 , 16, $28 - 35.$

(44) Reasoner, D. J.; Geldreich, E. E. Appl. Environ. Microbiol. 1985 , 49, 1 −7.

(45) Maldonado, L. A.; Fragoso-Yáñez, D.; Pérez-García, A.; ́ ∫
∫ Rosellón-Druker, J.; Quintana, E. T. Antonie van Leeuwenhoek 2009, , 95, 111 −120.

(46) National Committee for Clinical Laboratory Standards. Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria that Grow Aerobically, 7th ed.; NCCLS: Villanova, PA, 2006; Approved standard M7 −A7.

(47) Marfey, P. Carlsberg Res. Commun. 1984 , 49, 591 −596.